

# DEVI AHILYA VISHWAVIDYALAYA, INDORE

## M.Sc. CHEMISTRY (SEMESTER -III)

Paper No.  
Compulsory / Optional  
Max. Marks

: I (Code-MCH-501 )  
: Compulsory  
: 100

### PAPER I : APPLICATION OF SPECTROSCOPY-I

<b>Unit - 1</b>	<b>Electronic Spectroscopy:</b> Electronic Spectral Studies for $d^1 - d^9$ systems in octahedral, tetrahedral and square planer complexes
<b>Unit - 2</b>	<b>Vibrational Spectroscopy</b> Symmetry and shapes of $AB_2$ , $AB_3$ , $AB_4$ , $AB_5$ and $AB_6$ , mode of bonding of ambidentate ligands, nitrosyl, ethylenediamine and diketonato complexes, application of resonance Raman spectroscopy and its applications.
<b>Unit - 3</b>	<b>Nuclear Magnetic Resonance Spectroscopy-I</b> General introduction and definition, chemical shift, spin-spin interaction, shielding and deshielding mechanism, mechanism of measurement of chemical shift values and correlation for protons bonded to carbon (aliphatic, olefinic, aldehydic and aromatic) and other nuclei (alcohols, phenols, enols, carboxylic acids, amines, amides & mercapto),
<b>Unit - 4</b>	<b>Nuclear Magnetic Resonance Spectroscopy-II</b> Chemical exchange, effect of deuteration, Complex spin spin interaction between two, three, four and five nuclei (I order spectra) Stereochemistry, hindered rotation, Karplus curve-variation of coupling constant with disordered angle. NMR shift reagents, solvent effects. nuclear overhauser effect (NOE).
<b>Unit - 5</b>	<b>Mössbauer Spectroscopy</b> Basic principles, spectral parameters and spectrum display. Application of the technique to the studies of (1) bonding and structures of $Fe^{+2}$ and $Fe^{+3}$ compounds including those of intermediate spin, (2) $Sn^{+2}$ and $Sn^{+4}$ compounds nature of M-L bond, coordination number, structure and (3) detection of oxidation state and inequivalent MB atoms.

A collection of handwritten signatures and dates, all written in black ink. The dates are consistently '2-9-16'. The signatures include:
 

- A circular signature with 'Omar' and '2-9-16' below it.
- A signature with 'Sharma' and '2-9-16' below it.
- A signature with 'HPS' and '2-9-16' above it, followed by 'HOS' and '2-9-2016' below it.
- A signature with 'S' and '2-9-16' below it.
- A signature with 'DML' and '2-9-16' below it.
- A signature with 'SPK' and '2-9-16' below it.
- A signature with 'Y. Mij' and '2-9-16' below it.
- A signature with 'P. D.' and '2-9-16' below it.
- A small circle with a small mark inside it.
- A small rectangular stamp.

- Structural Methods for Chemistry, R.S. Drago, Saunders Compnay.
- Structural Methods in Inorganic Chemistry, E.A.V. Ebsworth, D.W.H. Rankin and S. Cradock, ELBS.
3. Infrared and Raman Spectral : Inorganic and Coordination Compounds K. Nakamoto, Wiley.
  4. Progress in Inorganic Chemistry vol., 8, ed., F.A. Cotton, vol., 15 ed. S.J. Lippard, Wiley.
  5. Transition Metal Chemistry ed. R.L. Carlin vol. 3 dekker.
  6. Inorganic Electronic Spectroscopy, A.P.B. Lever, Elsevier.
  7. NMR, NQR, EPR and Mossbauer Spectroscopy in Inorganic Chemistry, .V. Parish, Ellis Haywood.
  8. Practical NMR Spectroscopy, M.L. Martin. J.J. Deepish and G.J. Martin, Heyden.
  9. Spectrometric Identification of Organic Compounds, R.M. Silverstein, G.C. Bassler adn T.C. Morrill, John Wiley.
  10. Introduction to NMR spectroscopy, R.J. Abraham, J. Fisher and P. Loftus, Wiley.
  11. Application of Spectroscopy of Organic Compounds, J.R. Dyer Prentice Hall.
  12. Spectroscopic Methods in Organic Chemistry D.H. Williams, I. Fleming, Tata McGraw-Hill.
  13. Structural Methods in Inorganic Chemistry, E.A.V. Ebsworth, D.W.H. Rankin and S. Cradock, ELBS.
  14. Introduction to NMR spectroscopy, R.J. Abraham, J. Fisher and P. Loftus, Wiley.

Q Shanna  
2-9-2016

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DKS 2-9-16

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AB 2-9-16

John 2-9-16