

STRY (SEMESTER -III)

: I (Code-MCH-503)

: Compulsory

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PAPER III: ENVIRONMENTAL CHEMISTRY

Atmospheric layers, Vertical temperature profile, heat/radiation budget of the ea atmosphere systems. Properties of troposphere, thermodynamic derivation of lapse r Temperature inversion. Calculation of Global mean temperature of the atmospher Pressure variation in atmosphere and scale height. Biogeochemical cycles of carb nitrogen, sulphur, phosphorus, oxygen. Residence times. Atmospheric Chemistry Sources of trace atmospheric constituents: nitrogen oxides, sulphurdioxide and of sulphur compounds, carbon oxides, chlorofluorocarbons and other halogen compoun methane and other hydrocarbons. Tropospheric Photochemistry Mechanism of Photochemical decomposition of NO2 and formation of ozone. Format of oxygen atoms, hydroxyl, hydroperoxy and organic radicals and hydrogen peroxic Reactions of hydroxyl radicals with methane and other organic compounds. Reaction OH radicals with SO2 and NO2. Formation of Nitrate radical and its reaction Photochemical smog meteorological conditions and chemistry of its formation. Unit-2 Air Pollution Air pollutants and their classifications. Aerosols-sources, size distribution and effect on visibility, climate and health. Acid Rain Definition, Acid rain precursors and their aqueous and gas phase atmospheric oxidation reactions. Damaging effects on aquatic life, plants, buildings and health. Monitoring of SO2 and NO2. Acid rain control strategies. Stratospheric Ozone Depletion Mechanism of Ozone formation, Mechanism of catalytic ozone depletion, Discovery of Antarctic Ozone hole and Role of chemistry and meteorology. Control Strategies. Green House Effect Terrestrial and solar radiation Spectra, Major green house gases and their sources and Global warming potentials. Climate change and consequences. Urban Air Pollution Exhaust emissions, damaging effects of carbon monoxide. Monitoring of CO. Control strategies. Unit-3 Aquatic Chemistry and Water Pollution Redox chemistry and waters. Dissolved oxygen, biological oxygen demand, chemical oxygen demand, determination of DO, BOD and	l mrt-1	Atmosphere
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Bioaccumulation, sources of heavy metals. Chemical speciation of Hg, Pb, As, and Cd. Biochemical and damaging effects.	Unit-4	Environmental Toxicology Toxic heavy metals: Mercury, lead, arsenic and cadmium. Causes of toxicity. Bioaccumulation, sources of heavy metals. Chemical speciation of Hg. Ph. As, and Cd.

	Application of the control of the co
	Organic Compound: Pesticides, classification, properties and uses of
The state of the s	reamochlorine and ionospheres pesticides detection and damaging effects.
- Address of the second	Polychlorinated biphenyls: Properties, use and environmental continuation and effects.
	Polynuclear Aromatic Hydrocarbons: Source, structures and as pollutants.
Unit-5	Soil and Environmental Disasters
	Soil composition, micro and macronutrients, soil pollution by fertilizers, plastic an metals.
	Methods of re-mediation of soil. Bhopal gas tragedy, Chernobyl, three mile island,
	Minimtata Disease, Sevoso (Italy), London smog.

Books Suggested:

- 1. Environmental Chemistry, Colin Baird, W.H. Freeman Co. New York, 1998.
- 2. Chemistry of Atmospheres, R.P. Wayne, Oxford.
- 3. Environment Chemistry, A.K. De, Wiley Eastern, 2004.
- 4. Environmental Chemistry, S.E. Manahan, Lewis Publishers.
- 5. Introduction to atmospheric Chemistry, P.V. Hobbs, Cambridge.

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